RESPONSE OF SOIL PHOSPHATASES TO THREE DIFFERENT IONIC LIQUIDS WITH HEXAFLUOROPHOSPHATE ANION

Arkadiusz Telesiński¹, Martyna Śnioszek¹, Robert Biczak², Barbara Pawłowska²

- ¹ Department of Plant Physiology and Biochemistry, West Pomeranian University of Technology, Słowackiego 17, 71-434 Szczecin, Poland, e-mail: Arkadiusz.Telesinski@zut.edu.pl
- ² Department of Biochemistry and Ecotoxicology, Jan Długosz University in Częstochowa, 13/15 Armii Krajowej Av., 42-200 Częstochowa, Poland

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ABSTRACT

The aim of this study was to determine the effect of three different ionic liquids (ILs): 1-butyl-1-methylpyrrolidynium hexafluorophosphate [Pyrrol][PF6], 1-butyl-1-methylpiperydynium hexafluorophosphate [Piper][PF6] and 1-butyl-4-methylopirydynium hexafluorophosphate [Piryd][PF6] on phosphatase activities in soil. The pot experiment was carried out on loamy sand (Corg content 9.0 g/kg) with spring barley as a tested plant. The ILs was used at the dosages of 0, 1, 10, 50, 100, 400, 700 and 1000 mg/kg dry matter (DM). On day 14, soil samples were collected and activities of alkaline phosphomonoesterase, acid phosphomonoesterase, phosphotriesterase and inorganic pyrophosphatase were determined. Based on the analysis of the effect measure n² by variance analysis - ANOVA - the percentage shares of all variable factors affecting the activity of phosphatases were also defined. Obtained results showed that the application of different dosages of all ILs caused mainly non-significant changes in phosphatase activity. Changes in activity of phosphatases were often not dependent on IL dosages. The most sensitive for ILs were both phosphomonoesterases. The type of ILs had the highest percentage participation in formation of acid phosphomonoesterase activity, while dosages of ILs affected mainly alkaline phosphomonoesterase and phosphotriesterase activities.

Keywords: ionic liquids, soil, phosphomonoesterases, phosphotriesterase, inorganic pyrophosphatase, resistance index

INTRODUCTION

In the search for environmentally friendly chemicals, especially solvents, the world of science has focused on ionic liquids – ILs [Biczak et al., 2016a]. Ionic liquids compose a group of chemical compounds which have an ionic nature, they consist an organic cation and inorganic or organic anion, and their melting point does not exceed 100°C. These compounds play a very important role in scientific investigations as well as in industrial organic synthesis, both in high-tonnage productions as well as low-tonnage technologies of high added-value chemicals and materials. Ionic liquids are mostly used as alternative media and/or catalysts for desired chemical reactions [Borowiecki et al., 2015]. Due to the very low vapor pressure, these compounds are often referred to as environment-friendly "green solvents" [Romero et al., 2008]. However the total perception of this compounds as environmentfriendly may lead to their improper experimental design and utilization. Important is also the fact that potential ecotoxity and existence of majority of ILs present in the environment have not been fully understood yet [Telesiński and Sułkowska, 2016], nevertheless the toxic effects of ILs on microorganisms, plants, aquatic organisms and soil animals have been recently evaluated [Docherty and Kulpa, 2005; Dong et al., 2013; Li et al., 2010; Liu et al., 2015; Biczak, 2016]. Due to the fact that ILs possess some excellent properties, such as considerable solubility in water and resistance to degradation, their environmental risks may enhance together with increasing scale of the ILs usage [Sun et al., 2017].

The ILs may enter the soil, among other things, as a result of accidental spillage, use of compost from sewage sludge or landfill leachate [Telesiński and Sułkowska, 2016]. According to Bielińska et al. [2005] phosphatases are the most frequent measured enzymes in soil affected with anthropogenic contamination. Activity of enzymes is one of the best indicators of any changes in soil environment caused by natural and anthropogenic factors, especially organic xenobiotics. Moreover changes in soil enzymatic activity are one of the earliest signals indicating some fluctuations in the intensity of environmental life processes [Bielińska and Mocek-Płóciniak, 2010].

The name phosphatase describes a group of enzymes that hydrolyses esters as well as anhydrides of phosphoric acid. There are different phosphatases in soils: phosphomonoesterase, phosphodiesterase, phosphotriesterase, phospholipase, inorganic pyrophosphatase and enzymes acting on P-N-bonds [Wang et al., 2011].

This paper presents the effect of three different ionic liquids with hexafluorophosphate anion on the activities of alkaline phosphomonoesterase, acid phosphomonoesterase, phosphotriesterase and inorganic pyrophosphatase in soil.

MATERIAL AND METHODS

The ionic liquids, such as: 1-butyl-1-methylpyrrolidynium hexafluorophosphate [Pyrrol] [PF6], 1-butyl-1-methylopiperydynium hexafluorophosphate [Piper][PF6] and 1-butyl-4-methylpirydynium hexafluorophosphate [Piryd][PF6] were used in the experiment.

A pot experiment was carried out in the vegetation hall of the Department of Biochemistry and Ecotoxicology at Jan Długosz University in Częstochowa. A monocotyledonous plant, the spring barley (*Hordeum vulgare* L.) was used in the experiment. Fifteen seeds of the plants, originating from the same source, were sown into a 90 mm-diameter plastic plant pot that was filled with the reference soil and a soil thoroughly mixed with the studied ILs. The ILs was used at the dosages of 0, 1, 10, 50, 100, 400, 700 and 1000 mg/kg dry matter (DM). The soil used in the experiment was loamy sand with a dissolved matter containing of approximately 10%, an organic carbon of 9.0 g/kg and pH in 1 M KCl equal to 6.0. Throughout the testing period (14 days), constant substrate moisture content at the level required for the plants (70% field water capacity), a constant temperature $20 \pm 2^{\circ}$ C and a constant illumination of 170 µmol/m²/s were maintained in the system of 16 h on day and 8 h on night.

On day 14 the soil samples were collected to analyze the activity of phosphatases. The acid phosphomonoesterase - Pac (EC 3.1.3.2) and alkaline phosphomonoesterase - Pal (EC 3.1.3.1) activities were determined as described by Tabatabai and Bremner [1969]. The phosphotriesterase - PT (EC 3.1.8.1) activity was determined according to Eivazi and Tabatabai [1977]. The yellow-band absorbance of the filtrate due to pnitrophenol was measured at 400 nm. The inorganic pyrophosphatase - IPP (EC 3.1.6.1) activity was measured by Dick and Tabatabai [1978] method. Released orthophosphate was extracted with sulfuric acid, and determined photometrically at 700 nm after colorization with ammonium molybdate. Enzyme activities were calculated using a calibration curve. Shimadzu UV-VIS 1800 equipment was used in all measurements.

Phosphatase activity determinations were performed on three replicates for each treatment, and the significance of the observed differences was verified using a one-way analysis of variance followed by the post-hoc Tukey's HSD test. Differences with a p value of < 0.05 were considered as significant.

Additionally, mean activity of phosphatase of samples treated with all dosages of ionic liquid was used to calculate the index of enzymatic resistance (RS) according to the formula developed by Orwin and Wardle [2004]:

$$\mathbf{RS} = 1 - \frac{2|\mathbf{D}|}{\mathbf{C} + |\mathbf{D}|} \tag{1}$$

where *C* is the soil resistance under natural conditions (not exposed to ILs) and P the resistance of soil exposed to ILs; D = C-P.

Based on the analysis of the effect measure η^2 by variance analysis – ANOVA – the percentage shares of all variable factors affecting the activity of phosphatases were also defined.

RESULTS AND DISCUSSION

Activity of alkaline phosphomonoesterase (Pal), acid phosphomonoesterase (Pac), phosphotriesterase (PT) and inorganic pyrophosphatase (IPP) in control soil (without ILs) was as follows: 174.54–186.02 mg p-NP/kg DM/h, 116.26–145.76 mg p-NP/kg DM/h, 22.22–23.26 mg p-NP/kg DM/h and 406.34–487.71 mg P-PO₄³⁻/kg DM/h, respectively.

Application of different dosages of [Pyrrol] [PF6] caused mainly non-significant changes in activity of all phosphatases. Significant decrease in activity of Pal was observed only in soil treated with [Pyrrol][PF6] at the dosage of 50 mg/kg DM (30.71% compared to control). Inhibition of Pac was also reported in soil containing [Pyrrol][PF6] at dosages: 700 and 1000 mg/kg DM (35.04 and 41.19% compared to control, respectively). Additionally, this IL at the dosage of 1000 mg/kg decreased in IPP activity (24.80% compared to control). However in soil with [Pyrrol][PF6] at the dosage of 10 mg/kg DM a stimulation of IPP was noted (25.89% compared to control) (Table 1).

Similarly to [Pyrrol][PF6], after treatment with [Piper][PF6] mostly non-significant changes in activity of PT and also IPP were observed. Significant increase of IPP activity appeared only in soil with addition of [Piper][PF6] at the dosage of 1 mg/kg DM (56.11% compared to the control). Whereas, application of this IL mainly significantly affected Pal and Pac activities. In the soil containing [Piper][PF6] at dosages of: 1, 50 and 100 mg/kg DM a significant stimulation of Pal activity was noticed (16.85, 15.09 and 23.86% compared to the control, respectively). However a significant increase in Pal activity was reported in soil treated with [Piper][PF6] at dosages of 10, 400, 700 and 1000 mg/kg DM (in the range from 19.95 to 36.40% compared to the control). Additionally, significant inhibition of Pac occurred after application of this IL at dosages: 10, 50, 400, 700 and 1000 (in the range from 19.72 to 49.79% compared to the control) (Table 2).

Application of [Piryd][PF6] at the dosages of 1–50 mg/kg DM did not affect significantly activity of all determined phosphatases (Table 3). This effect was also observed for PT in soil treated with all dosages of [Piryd][PF6]. In soil with [Piryd][PF6] at the dosage of 100 mg/kg DM a significant decrease in enzyme activity was observed only for IPP (9.75% compared to control). Application of this IL at the dosages of 400 and 700 mg/kg DM caused a significant inhibition only

ILs dosage	Pal	Pac	PT	IPP
[mg/kg DM]	[mg p-NP/kg DM/h]	[mg p-NP/kg DM/h]	[mg p-NP/kg DM/h]	[mg P-PO ₄ ³⁻ /kg DM/h]
0	179.54 bc	145.76 a	23.26 a	487.71 bc
1	215.95 a	147.48 a	23.26 a	497.25 bc
10	215.24 a	144.13 a	24.97 a	613.99 a
50	124.40 d	118.03 ab	23.10 a	535.95 b
100	208.25 ab	130.87 a	23.22 a	469.37 c
400	162.47 bc	135.55 a	23.16 a	541.73 b
700	142.49 cd	94.68 bc	22.21 a	554.71 b
1000	153.14 bcd	85.72 c	21.41 a	366.77 d

Table 1. Activity of soil phosphatases in soil treated with different dosages of [Pyrrol][PF6]

Values denoted with the same letter for each enzyme do not differ statistically at p < 0.05

Table 2. Activity of soil phosphatases in soil treated with different dosages of [Piper][PF6]

ILs dosage [mg/kg DM]	Pal [mg p-NP/kg DM/h]	Pac [mg p-NP/kg DM/h]	PT [ma p-NP/ka DM/h]	IPP [ma P-PO. ³⁻ /ka DM/h]	
0	174.30 b	143.63 a	22.22 a	406.34 b	
1	203.68 a	146.23 a	23.23 a	634.35 a	
10	110.86 d	94.96 c	24.76 a	455.74 b	
50	200.61 a	115.31 b	21.93 a	417.51 b	
100	215.89 a	129.99 ab	21.49 a	481.87 b	
400	139.53 c	85.90 cd	24.43 a	442.41 b	
700	128.85 cd	79.47 cd	23.70 a	427.28 b	
1000	124.87 cd	72.11 d	19.25 a	487.97 b	

Values denoted with the same letter for each enzyme do not differ statistically at p < 0.05

ILs dosage [mg/kg DM]	Pal [mg p-NP/kg DM/h]	Pac [mg p-NP/kg DM/h]	PT [mg.p-NP/kg.DM/h]	IPP [mg P-PO, ³⁻ /kg DM/h]
0	186.02 ab	116.26 a	22.25 a	452.28 b
1	199.71 a	111.00 a	25.05 a	431.94 bc
10	178.93 ab	96.73 ab	26.05 a	506.83 ab
50	158.25 bc	96.73 ab	21.26 a	497.15 b
100	150.30 bcd	96.06 ab	18.78 a	408.18 c
400	129.89 cde	85.90 ab	19.12 a	469.99 b
700	110.11 de	79.47 ab	21.14 a	453.85 b
1000	115.43 e	72.11 b	20.01 a	568.42 a

Table 3. Activity of soil phosphatases in soil treated with different dosages of [Piryd][PF6]

Values denoted with the same letter for each enzyme do not differ statistically at p < 0.05

of Pal activity (30.17 and 40.81% compared to the control, respectively). However treating soil with the highest dosage (1000 mg/kg DM) of [Piryd][PF6] resulted in significant decrease in Pal and Pac activity (37.95 and 37.97% compared to control, respectively) and an increase IPP activity (25.68% compared to control).

An interpretation of the effect of different ILs and any other pollutant on soil phosphatase activities is difficult because the enzyme assays currently used have some limitations and because of the existence of direct and indirect effects on the measured enzyme [Nannipieri et al., 2003]. For example, inhibition of enzyme activity by a pollutant may be masked by growth of surviving microorganisms which are characterised with expression of genes codifying the enzyme. A relative increase of biomass is a result of growing population of microorganisms which survive a toxic treatment and may feed on microbial remains of organisms which do not survive it [Nannipieri et al., 2011]. According to Griffiths and Philippot [2013] the index of enzymatic resistance values

can be used to determine the sensitivity of analysed ecosystem to various stressors. Soil resistance indicators provide information about the status of soil environments contaminated with organic compounds [Kaczyńska et al., 2015]. Additionally Parelho et al. [2016] reported that soil phosphatases are important factors in the process of organic phosphorus compounds transformation to inorganic phosphates, and they are very often used as the indicator of soil fertility.

It was found, based on the calculated values of resistance index (RS), that among all phosphatases, the most sensitive to the ILs analysis in soil were Pal and Pac. The RS values for the soil treated with [Pyrrol][PF₆], [Piper][PF6] and [Piryd][PF6], in case of Pal, were 0.67, 0.54 and 0.43, respectively, and in case of Pac 0.67, 0.56 and 0.57, respectively (Fig. 1). The values of RS calculated for PT and IPP were significantly higher, in the range of 0.79–0.94 and 0.77–0.85, respectively.

Results obtained by Sun et al. [2017] revealed that acid phosphatase, as well as dehydrogenase, was not sensitive to 1-methyl-3-octylimidazoli-



Figure 1. Mean resistance indices of phosphatase activitis in soil treated with [Pyrrol][PF6], [Piper][PF6] and [Piryd][PF6]

um tetrafluoroborate, while urease activity was first stimulated and then inhibited by this IL (observed after 30-day incubation). Telesiński and Sułkowska [2016] reported that two imidazolium ionic liquids: 3-butyl-1-methylimidazolium tetrafluoroborate and 3-hexyl-1-methylimidazolium tetrafluoroborate, decreased activity of o-diphenol oxidase, but the size of this effect depended on a dosage of ILs, incubation time and soil properties. However, Li et al. [2013] showed that imidazolium ionic liquids may improve enzyme activities because they participate in biocatalytic processes.

The data presented in Table 4 indicated unequivocally that the activity of phosphatases varied depending on the type and the dosages of ILs. The share of types of IL took the major role in the formation of Pac activity (56.67%). However, the activity of Pal and PT was mainly formed by IL dosages (57.33 and 50.69%, respectively). In contrast, for IPP activity the highest impact was observed in case of interaction between both: a type and a dosage of ILs. Our previous studies showed that effect of different ionic liquids on biochemical parameters of soil and plants largely depended on the structure of the examined ILs cations, IL dosages and type of anion [Biczak, 2016, Biczak et al., 2014, 2016a, 2016b, Telesiński 2012, Telesiński and Sułkowska 2016].

CONCLUSIONS

In the present study, the effects of three different ionic liquids with hexafluorophosphate anion: [Pyrrol][PF6], [Piper][PF6] and [Piryd] [PF6] on soil phosphatase activities were examined. First, the application of different dosages of all ILs caused mainly non-significant changes in all phosphatase activities. Secondly, changes in phosphatase activities were often not dependent on IL dosages, but the most sensitive for ILs were phosphomonoesterases. Thirdly, type of ILs affected mainly activity of acid phosphomonoesterase, while dosages of ILs activity of alkaline phosphomonoesterase and phosphotriesterase. The result of the presented study is beneficial for future research about ILs toxicity.

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Table 4. Participation of variable factors in the formation of phosphatase activities [%]

Variable factors	Pal	Pac	PT	IPP
Type of ILs (A)	22.26	56.67	26.18	29.86
Dosage of ILs (B)	57.33	39.04	50.69	18.75
A×B	19.00	3.61	13.72	43.59
Error	1.41	0.68	9.41	7.81

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